Partial B-to-A DNA Transition upon Minor Groove Binding of Protein Sac7d Monitored by Raman Spectroscopy[†]

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ABSTRACT: Members of the Sso7d/Sac7d protein family and other related proteins are believed to play an important role in DNA packaging and maintenance in archeons. Sso7d/Sac7d are small, abundant, basic, and nonspecific DNA-binding proteins of the hyperthermophilic archeon Sulfolobus. Structures of several complexes of Sso7d/Sac7d with DNA octamers are known. These structures are characterized by sequence unspecific minor groove binding of the proteins and sharp kinking of the double helix. Corresponding Raman vibrational signatures have been identified in this study. A Raman spectroscopic analysis of Sac7d binding to the oligonucleotide decamer d(GAGGCGCCTC)₂ reveals large conformational perturbations in the DNA structure upon complex formation. Perturbed Raman bands are associated with the vibrational modes of the sugar phosphate backbone and frequency shifts of bands assigned to nucleoside vibrations. Large changes in the DNA backbone and partial B- to A-form DNA transitions are indicated that are closely associated with C2'-endo/anti to C3'-endo/anti conversion of the deoxyadenosyl moiety upon Sac7d binding. The major spectral feature of Sac7d binding is kinking of the DNA. Raman markers of minor groove binding do not largely contribute to spectral differences; however, clear indications for minor groove binding come from G-N2 and G-N3 signals that are supported by Trp24 features. Trp24 is the only tryptophan present in Sac7d and binds to guanine N3, as has been demonstrated clearly in X-ray structures of Sac7d-DNA complexes. No changes of the Sac7d secondary structure have been detected upon DNA binding.

Complex formation with DNA-binding proteins drives DNA superhelicity, which in turn regulates the double helix stability. DNA in eubacteria and eukarya is overall negatively supercoiled due to gyrase action and wrapped around the histone core of the nucleosome. In contrast, episomal DNAs in thermophilic archaea are relaxed to positively supercoiled forms (1, 2). In archaea the mechanism for organizing DNA into a compact form is not well understood. Members of the Sso7d/Sac7d protein family and other related proteins are believed to play an important role in DNA packaging and maintenance. Sso7d/Sac7d are small, abundant, basic, and nonspecific DNA-binding proteins of the hyperthermophilic archeon *Sulfolobus* (3, 4).

Sso7d/Sac7d bind to DNA noncooperatively with micromolar affinity, without strong sequence preference, and the $T_{\rm m}$ values of Sac7–DNA complexes are \sim 40 °C higher than those of free DNA (5). The structures of several Sac7d/Sso7d–DNA complexes have been studied by X-ray crystallography (6–8), NMR (9), and low-angle X-ray scattering

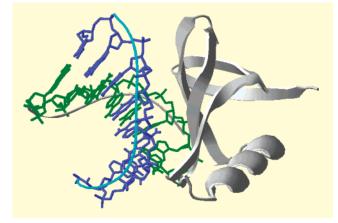


FIGURE 1: Model of the Sac7d-d(GCGATCGC)₂ complex. The figure was prepared using SPDBV (DeepView/Swiss-PdbViewer from http://www.expasy.org/spdb/) and the atomic coordinates from the pdb file 1AZP.

(10). Figure 1 shows a model of the Sac7d-DNA complex with octamer d(GCGATCGC)₂ (6).

In the protein—DNA complexes, the protein structures are similar to those of the free proteins (3, 11), consisting of an incomplete β -barrel made of a triple-stranded β -sheet orthogonal to a β -hairpin. The small β -barrel is capped by an amphiphilic C-terminal α -helix. The triple-stranded β -sheet is placed across the DNA minor groove, and the side chains of Val26 and Met29 intercalate into DNA base pairs

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and cause a sharp kink (\sim 60°) in the DNA duplex (6-8).

In the present study, we analyze by Raman spectroscopy the binding of Sac7d to the 10-bp DNA oligonucleotide d(GAGGCGCCTC)₂. A Raman difference spectrum was obtained by subtraction of the spectra of isolated Sac7d and d(GAGGCGCCTC)₂ from the spectrum of the Sac7d—d(GAGGCGCCTC)₂ complex, and its analysis reveals large conformational perturbations in the DNA structure induced by Sac7d protein binding.

MATERIALS AND METHODS

Chemicals. Isopropyl- β -D-thiogalactopyranoside (IPTG),¹ ampicillin, and chloramphenicol were obtained from MdBio, Inc., Taipei, Taiwan. All other chemicals were purchased from Merck, Darmstadt, Germany. Superdex 200 was from Amersham-Pharmacia, Freiburg, Germany.

Preparation of Sac7d, $d(GAGGCGCCTC)_2$, and the Sac7d– $d(GAGGCGCCTC)_2$ Complex for Raman Measurements. Recombinant Sac7d was overexpressed in *E. coli*, after induction with IPTG, and purified as described elsewhere (4). The purified protein was dialyzed against deionized water and lyophylized. Before Raman measurements the Sac7d protein was dissolved in 50 mM Tris-HCl pH 7.5, 50 mM NaCl, and 10 mM MgCl₂ (buffer I) and filtered through a Superdex 200 column, equilibrated with buffer I. The protein samples were concentrated in Centricon YM-3, 3000 MWCO Millipore tubes. Protein concentrations were determined by UV absorbance using an extinction coefficient of $\epsilon_{278\text{nm}} = 8400 \text{ M}^{-1} \cdot \text{cm}^{-1}$ calculated from the amino acid sequence with the Expasy server.

Oligonucleotides were purchased from BioTez, Berlin, Germany. Concentrations of DNA were determined spectrophotometrically using an extinction coefficient $\epsilon_{260\text{nm}} = 14\ 120\ \text{M}\cdot\text{bp}^{-1}\cdot\text{cm}^{-1}$ calculated from the base composition according to ref 12. Equimolar amounts of each strand were mixed, heated to 95 °C for 10 min, annealed by slow cooling, and purified by gel filtration using a Superdex 200 column equilibrated in buffer I. Peak fractions were concentrated in Centricon Millipore tubes YM-3, 3000 MWCO.

The Sac7d-d(GAGGCGCCTC)₂ complex was prepared by mixing Sac7d and d(GAGGCGCCTC)₂ in a 1:1 molar ratio. The samples were concentrated in a 5K Millipore Ultrafree centrifugal filter device. The concentration of the complex was determined from the absorbance at 260 nm using the DNA extinction coefficient, neglecting the minor contribution of protein to the absorbance.

Hydrogen−*Deuterium Exchange*. Hydrogen−deuterium exchange was accomplished by incubation of folded Sac7d in D₂O based buffer II (50 mM Tris-DCl, pD 7.5, 50 mM NaCl, and 10 mM MgCl₂) and passing the samples through a homemade Sephadex G10 (Pharmacia) column of 2 cm length. The column was equilibrated in D₂O with buffer II. Protein samples elute in the void volume and were fully separated from the H₂O solvent with buffer I. This procedure accomplishes only H → D exchange (OH → OD and NH → ND) of hydrogen atoms that are not protected by covalent bonds or the native fold of the protein.

Raman Spectroscopy. Homemade quartz cuvettes were filled with approximately 12 μ L of the sample solution and

then tightly closed with Teflon stoppers. The sample concentrations in H₂O buffer I (the concentrations in D₂O buffer II are given in parentheses) were 11 and 12 (13) mg/ mL for DNA, 12 and 22 (17) mg/mL for Sac7d, and 11 and 50 (36) mg/mL for the Sac7d-d(GAGGCGCCTC)₂ complex. Spectra were collected at 22 °C on the Raman spectrometer T64000 (Jobin Yvon, France) in the single monoconfiguration equipped with a liquid nitrogen-cooled chargecoupled-device (CCD) detector. The 488-nm line of a Coherent Innova 90 argon ion laser with 200 mW of radiant power at the sample position in the macrochamber was used for the excitation of the Raman spectra. Ten spectra, each measured for 300 s, were accumulated and averaged to produce the spectra shown in Figures 2-4. To exclude all possible drifts of the wavenumber scale during the data measurement, a calibration spectrum was collected after each 300 s sample spectra accumulation step. The calibration procedure is described in detail elsewhere (13).

Raman data analyses were performed with the software packages LabSpec (Jobin Yvon) and GRAMS (Thermo Galactic). Minimization of intensity differences between the spectra of isolated components and Sac7d-d(GAGGCGC-CTC)₂ is crucial for the calculation of difference spectra. To achieve this, the d(GAGGCGCCTC)₂ spectrum was normalized with respect to the P-O stretching vibration of the phosphodioxy group (PO₂⁻) near 1092 cm⁻¹, and the Sac7d spectrum was normalized with respect to the phenyl ring vibration of phenylalanine at 1003 cm⁻¹. Normalized but otherwise not corrected experimental spectra of d(GAG-GCGCCTC)₂ and Sac7d were subtracted from the spectrum of the Sac7d-d(GAGGCGCCTC)₂ complex. In the next evaluation steps, the buffer spectrum and fluorescence background were removed. For this purpose the fluorescence background was approximated by a polynomial curve.

Difference bands were considered as significant when the following criteria are fulfilled: (i) the intensity of the difference band is at least two times higher than the signal-to-noise ratio, (ii) the difference band reflects an intensity change of at least 5% of its parent band, and (iii) corresponding difference bands were obtained from spectra measured in H_2O and D_2O buffers.

RESULTS

Raman Spectrum of d(GAGGCGCCTC)₂

The self-complementary DNA sequence of the decameric oligonucleotide d(GAGGCGCCTC)₂ used in this study is closely related to that of the octamers studied previously in crystal structure analyses of Sac7d-DNA complexes (6–8). Figure 2 shows the Raman spectra of d(GAGGCGC-CTC)₂ in H₂O and D₂O buffers.

Wavenumber positions of the major peaks are given in Figure 2. The peak positions are in general accordance with those of DNA given previously in the literature (14-17 and references therein).

The backbone conformation markers at 831 and 1092 cm⁻¹ in H₂O and at 828 and 1091 cm⁻¹ in D₂O are diagnostic of B-DNA (*15*). The positions of the nucleoside conformation markers shift upon deuteration [from 667 to 661 cm⁻¹ (dT); 729 to 722 cm⁻¹ (dA); 749 to 747 cm⁻¹ (dT); 782 to 772 cm⁻¹ (dC); 1257 to 1261 cm⁻¹ (dC)]. The 683 cm⁻¹ dG marker band does not shift.

¹ Abbreviation: IPTG, isopropyl- β -D-thiogalactopyranoside.

FIGURE 2: Raman spectra of the 10-bp oligonucleotide $d(GAGGCGCCTC)_2$ (decamer I) in H_2O (2) and D_2O (1) solution. The sample buffer is 50 mM Tris-HCl pH 7.3 for spectrum 2 or pD 7.5 for spectrum 1 and 50 mM NaCl; data were collected at 22 °C in the region 590–1750 cm⁻¹. Concentrations are 12 mg/mL (H_2O) and 13 mg/mL (D_2O). Spectra were normalized with respect to the integrated intensity of the phosphate band near 1092 cm⁻¹ and the adenine-guanine band near 1580 cm⁻¹ (I7, II). Peak positions of prominent Raman bands are labeled, the given wavenumbers are accurate to within II cm⁻¹. Abbreviations are as follows: II G, cytosine; II A, adenine; II bk, deoxyribose backbone.

The nucleoside conformation markers identify C2′-endo/anti conformers. The spectra shown in Figure 2 are the signature of d(GAGGCGCCTC)₂ and provide the basis for the interpretation of the difference spectra shown in Figure 5.

Raman Signature of Sac7d

The Raman spectrum of Sac7d in H₂O (Figure 3, trace 2) is characterized by a strong amide I band (1640–1680 cm⁻¹) centered at 1670 cm⁻¹ and a broad, complex amide III band (1230-1310 cm⁻¹) with a major contribution at 1242 cm⁻¹ and weak shoulders at 1250 (irregular structures), 1268, 1285, and 1306 cm $^{-1}$ (α -helix). The features of the amide I and amide III bands indicate β -strands as the main secondary structure elements of Sac7d. Calculation of the Sac7d secondary structure on the basis of the amide I band intensity profile (18) gives $50 \pm 4\% \beta$ -strand, $15 \pm 4\% \alpha$ -helix, and $35 \pm 3\%$ irregular structures. The data are in agreement with the crystal structure of Sac7d (6). Figure 3 shows the Raman spectrum of Sac7d in fully protonated (nonexchanged) form measured in H₂O buffer (trace 2) and the Raman spectrum of the partially deuterated Sac7d obtained in D₂O (trace 1). Trace 3 shows the computed Raman difference spectrum (trace 1 minus trace 2) and represents the native exchange, that is, the exchange of hydrogen atoms that are exchangeable in the folded state of Sac7d. Information about the secondary

structure of natively exchanged and protected sites of Sac7d can be deduced from the large difference features at the positions of the amide I and amide III bands (in H_2O) and the positions of the amide I' and amide III' bands (in D_2O). The Raman intensity at the positions of the amide III and amide I bands is reduced and causes troughs in the difference spectrum.

Concomitantly, the Raman intensity at the positions of the amide III' and amide I' bands is increased and gives rise to difference peaks. Several trough/peak features are caused by the native state-deuteration induced frequency shift of amide III to amide III'. The trough/peak feature near 1239 and 986 cm⁻¹ is caused by β -strands, and that near 1251 and 926 cm⁻¹ is caused by irregular structures. Deuteration of the few α-helical secondary structures present in Sac7d results in complicated features at 1305, 1282, 1265, and 948 cm⁻¹. In the amide III region of Sac7d remain weak bands because of incomplete deuteration under the applied conditions of native exchange. The features of the amide I and amide I' bands with trough and peak, respectively, near 1674 and 1658 or 1640 cm⁻¹ provide additional Raman signatures for native (incomplete) deuterium exchange of secondary structure elements.

A very weak amide II band at 1550–1560 cm⁻¹ (Figure 3-2) represents N–H in-plane bending and C–N stretching of the trans peptide group. Deuteration of those groups causes

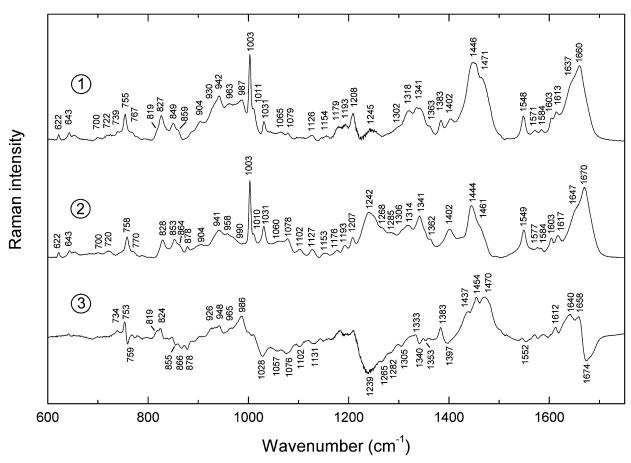


FIGURE 3: Raman spectra of Sac7d: (1) natively H → D exchanged Sac7d; (2) native Sac7d; (3) difference spectrum (spectrum 2 minus spectrum 1). Excitation at 488 nm; spectral region 600–1750 cm⁻¹; protein concentrations are 22 mg/mL (H₂O) and 17 mg/mL (D₂O); buffers are 50 mM Tris-HCl, pH 7.5 in H₂O and pD 7.5 in D₂O, and 50 mM NaCl; data collection at 22 °C. Spectra 1 and 2 were normalized to the integrated intensity of the phenylalanine band at 1003 cm⁻¹, which is not affected by deuteration (20). For the calculation of the difference spectrum (spectrum 3), the normalized spectra (spectra 1 and 2) were subtracted such that the intensity at the phenylalanine band 1003 cm⁻¹ becomes zero. Peaks in the difference spectrum (spectrum 3) appear when natively unprotected protons are exchanged against deuterons and the Raman bands are shifted. Peak positions of prominent Raman bands are labeled by the respective wavenumbers that are accurate to within ± 1 cm⁻¹.

an intense amide II' band at 1471 cm⁻¹ (Figure 3-1) and results in the most prominent feature of the difference spectrum with a peak at 1470 cm⁻¹ (Figure 3-3). This band can be assigned to non-α-helical (unprotected) sites of Sac7d (19).

Amino Acid Side Chains. The Raman spectra of Sac7d (Figure 3, traces 1 and 2) show typical contributions of aromatic side chains of one tryptophan, two phenylalanines, and two tyrosines and bands of nonaromatic amino acids (Table 1). Assignments were performed according to the literature (19-21) and references therein).

Environment of Tyrosine Side Chains. Two tyrosines, Y8 and Y34, are present in Sac7d. A tyrosine doublet at \sim 850 and $\sim 830 \text{ cm}^{-1}$ (about the same position in H₂O and D₂O) is caused by Fermi resonance of the normal mode v_1 (ring breathing fundamental) and the second harmonic $2v_{16a}$ (ring deformation overtone) of the para-substituted phenolic side chain (22).

The intensity ratio I_{850}/I_{830} is sensitive to hydrogen bonding of phenolic OH groups and, therefore, is an indicator of the tyrosine environment. In the H₂O spectrum (Figure 3-2) of Sac7d, two relatively sharp peaks of approximately equal intensity appear at 828 and 853 cm⁻¹. The intensity ratio $I_{853}/I_{828} = 1$ is in the range expected for tyrosines acting as both donor and acceptor of moderately strong or weak

hydrogen bonds, as is the case when they are exposed to solvent H_2O molecules (22). After $H \rightarrow D$ exchange (Figure 3-1) these bands marginally shift to 827 and 849 cm⁻¹, respectively, and change the intensity profiles, with the band at 827 cm⁻¹ becoming sharper and more intense than the band at 849 cm⁻¹.

Tryptophan. The Trp vibrations are sensitive to the indole ring environment. Buried residues form sharp intense peaks, whereas the intensity is low for exposed residues. The position of the indole ring vibration at 878 cm⁻¹ (called the W17 mode) of the unique Trp24 residue is indicative of exposed Trp residues with medium strong hydrogen bonds of the exocyclic Trp 1NH donors to water molecules. For the W17 mode, the frequency varies between 883 and 871 cm⁻¹; without H-bonding at the N1 site of Trp this band is located at 883 cm⁻¹, and with strong H-bonding the band shifts to 871 cm⁻¹ (23). In D₂O buffer the 878 cm⁻¹ band of Sac7d vanishes and a new 859 cm⁻¹ band appears. A similar downshift on deuteration was observed for free tryptophan (24).

The frequency of the Trp band near 1549 cm⁻¹ (called the W3 mode) assumes values between 1542 and 1557 cm⁻¹ depending on the absolute value of the $C_{\alpha}C_{\beta}-C_{3}C_{2}$ torsion angle $|\chi^{2,1}|$, which varies between 60° and 120° (25). The position of the W3 mode does not vary in the spectra of

FIGURE 4: Raman spectra of Sac7d, d(GAGGCGCCTC)₂, and the Sac7d-d(GAGGCGCCTC)₂ complex (Sac7d-decamer I): (A) protein in H₂O buffer; (B) protein in D₂O buffer (50 mM Tris-D₂O, pH or pD 7.5, and 50 mM NaCl). Protein concentrations are 50 mg/mL (H₂O) and 36 mg/mL (D₂O); spectra were excited at 488 nm, and data were collected at 22 °C in the region 600–1750 cm⁻¹. The blue traces show the Raman spectra of the isolated native and natively exchanged Sac7d from Figure 3. The green traces show the Raman spectra of decamer I from Figure 2. The red traces show the spectra of the Sac7d-d(GAGGCGCCTC)₂ complex. The black bottom traces show the computed difference spectra obtained by subtraction of the isolated component spectra from the experimental spectrum of the complex. The spectra are normalized to represent the same amounts of protein and DNA in the complex and in free form.

deuterated and nondeuterated samples, and its position of 1548 \pm 0.5 cm $^{-1}$ indicates an average $|\chi^{2,1}|$ value of ${\sim}88^{\circ}.$

The Fermi doublet of tryptophan with components at about 1360 and 1340 cm⁻¹ (with similar positions in the H₂O and D₂O spectra) serves as a hydrophobicity marker. The upper 1360 cm⁻¹ component of the doublet is strong in hydrophobic solvents whereas in a hydrophilic environment the lower 1340 cm⁻¹ component is stronger. The intensity ratio, R = I_{1360}/I_{1340} , is <0.9 for hydrophilic solvents and >1.1 for hydrophobic solvents (23). The intensity ratio is usually not detectable in proteins due to overlapping with bands caused by CH bending vibrations of aliphatic side chains. The Fermi doublet might be detected in difference spectra obtained by subtraction of the spectrum of the deuterated protein from that of the nondeuterated protein. Both bands of the doublet downshift by approximately 8 cm⁻¹ upon deuteration (23– 24) that might result in detectable features of the difference spectrum. In fact, a very weak difference band near 1353 cm⁻¹ and a relatively strong trough/peak feature at 1340/ 1333 cm⁻¹ are visible in Figure 3-3, resulting from the downshift of both Fermi doublet components. The ratio, R \approx 0.5, of the two bands of the Trp Fermi doublet is consistent

with the presence of a surface localized tryptophan indole ring being exposed to the water molecules of the buffer.

Several other tyrosine and aliphatic amino acid side chains contribute to the Sac7d Raman spectra shown in spectra 1 and 2 of Figure 3. The assignments of Raman bands and corresponding $H \rightarrow D$ exchange wavenumbers are listed in Table 1.

Raman Analysis of the $Sac7d-d(GAGGCGCCTC)_2$ Complex

The Raman spectra of the Sac7d—d(GAGGCGCCTC)₂ complex and its components are given in panels A and B of Figure 4. Panel A shows the Raman spectrum of fully protonated Sac7d, and panel B shows the spectrum of partially deuterated Sac7d. The blue, green, and red traces represent the spectra of Sac7d, d(GAGGCGCCTC)₂, and the complex, respectively, as given before in Figures 2 and 3. The black trace is the computed difference spectrum that was obtained by subtraction of the component spectra (green and blue traces) from the spectrum of the complex (red trace). Numerous peaks and troughs demonstrate spectral changes caused by complex formation between Sac7d and DNA. A

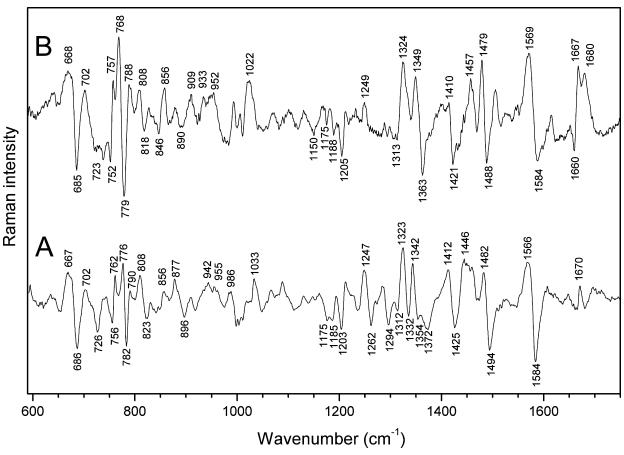


FIGURE 5: Enlarged Raman difference spectra (black bottom traces from Figure 4) of the Sac7d-decamer I complex in H₂O buffer (A) and in D₂O buffer (B). Spectra were smoothed using the 11-point Savitsky-Golay algorithm.

Table 1: Vibrational Assignments for Raman Markers of Sac7d in H₂O and D₂O Solutions^a

H_2O	D_2O	assignment b	H_2O	D_2O	assignment ^b
622	622	F	1153	1154	V, I
643	643	Y	1176	1179	$Y, [CH_3]$
654	653	?	1193	1193	?
700	700	M	1207	1208	F, Y
720	722	Y	1242	1245	AmIII (strand)
758	755	W	1268		AmIII (helix)
770	767	W	1285		AmIII (helix)
828	827	Y	1306	1302	[CH ₂], AmIII (helix)
853	849	Y	1314	1318	$[CH_2]$
878	859	W	1341	1341	$[CH_2]$
904	904	Q, S, T	1362	1363	W
	930	AmIII (coil, turns)		1383	W
941	942	K, V, L, AmIII (helix)	1402	1402	[COO ⁻]
958	963	K, L	1444	1446	K, I, L, [CH ₂]
	987	AmIII (strand)	1461		A, I, V, L, W, Y, [CH ₂]
990		I		1471	AmII
1003	1003	F	1549	1548	W
1010	1011	W	1577	1571	W
1031	1031	F	1584	1584	F
		K, A (C-C stretch), F			
1078	1079	C-C, C-N stretch	1617	1613	W
1102		P, A	1647	1637	AmI (helix)
1127	1126	I, V, L	1670	1660	AmI (strand, coil)

^a Raman data are from the Sac7d spectra shown in Figure 3. ^b Assignments to specific amino acid side chains (one letter abbreviations) are based upon literature data (19-21). Square brackets indicate chemical group frequencies common to more than one type of side chain. Amide (Am) modes are diagnostic of the indicated secondary structure.

positive difference peak indicates increased Raman intensity in the spectrum of the complex compared to the sum of component spectra, and a negative difference trough is caused by a lower intensity in the spectrum of the complex. When necessary, peak positions of deuterated samples will be given in parentheses.

Figure 5 shows the two amplified and, for clarity, smoothed difference spectra from Figure 4 (black traces) that are used for the further analysis. The features of the difference spectra provide information about conformational changes, structural rearrangements, and interactions between the components of the complex.

Backbone Conformation of d(GAGGCGCCTC)₂. The 700-900 cm⁻¹ region is characteristic for Raman markers of DNA backbone geometry (15). The 782 (772) cm⁻¹ peak in the Raman spectra of DNA is assigned to cytosine. The 791 (785) cm⁻¹ peak is assigned to a stretching vibration of backbone phosphodiester groups and diagnostic of B-form DNA backbone geometry, specifically of torsion angles α and ζ in the gauche⁻ range (26–28). The 782 cm⁻¹ cytosine and 791 cm⁻¹ backbone bands overlap in the H₂O spectrum of DNA to form a large 784 cm⁻¹ peak. In D₂O, the frequency of the cytosine peak changes to 772 cm⁻¹, and a broad band is formed with the backbone vibration that is indicated by a shoulder at 785 cm⁻¹. The difference spectra show a prominent peak/trough/peak feature that is probably composed of contributions from the DNA backbone band and the cytosine band. Changes in the DNA backbone band result in the peak at 790 (788) cm⁻¹ and the trough near 782 (779) cm⁻¹, and cytosine causes the peak at 776 (768) cm⁻¹ with the trough near 782 (779) cm⁻¹. The corresponding features in the D₂O difference spectrum (with band positions as given above in parentheses) support this assignment. The H_2O and D_2O difference features indicate small conformational perturbations to the C5'-O5'-P-O3'-C3' backbone network of DNA upon Sac7d binding. This is in agreement with previous observations (26, 29).

Deoxynucleoside Conformation of d(GAGGCGCCTC)₂. Sac7d binding alters the furanose ring geometry. The double stranded oligonucleotide d(GAGGCGCCTC)2 used in this study is composed of 80% GC and 20% AT. The difference spectra (Figure 5) indicate perturbations of the deoxynucleoside conformation of GC and AT base pairs. A very small adenosine band at 729 (722) cm⁻¹ is distorted under the influence of Sac7d binding. The intensity decrease of this band at 726 (723) cm⁻¹ and a peak/trough feature at 1342/ 1332 (Figure 5A) indicate perturbations at dA. Difference features with trough/peak near 685/668 cm⁻¹ and 1313/1324 cm⁻¹ (Figure 5B) are consistent with small changes at dG. Perturbations in cytosine nucleoside conformation markers are indicated in the difference spectra A (H₂O) and B (D₂O) by the respective peak/trough features at 776/782 (768/779) cm^{-1} and 1247/1262 (1249/-) cm^{-1} . Altogether, the observed spectral changes are consistent with partial transitions from C2'-endo/anti to C3'-endo/anti deoxynucleoside conformation at dG, dC, and dA. Similar shifts of those Raman markers accompany the canonical B- to A-form transition of DNA (30-32).

In DNA, for AT-rich sequences the phosphodiester B-form marker is located near 838–840 cm⁻¹, for GC-rich sequences the corresponding marker occurs near 828–830 cm⁻¹, and in the A-form DNA a phosphodiester marker appears near 807 cm⁻¹ (*14*). In the difference spectrum of the Sac7d–d(GAGGCGCCTC)₂ complex a prominent difference peak/trough feature at 808/823 cm⁻¹ and a small peak at 856 cm⁻¹ are indicative of a distortion of the DNA backbone and provide additional evidence for the backbone changes described above.

Deoxyribose Ring and Protein CH2/CH3 Vibrations. Furanose vibrations were identified in Raman spectra of DNA near 1420 and 1460 cm⁻¹ and assigned to backbone vibrations (26, 31, 33). In the spectrum of d(GAGGCGCCTC)₂ (Figure 2) the peak positions of those bands are at 1420 and 1461 cm⁻¹. Protein bands are also located in this spectral region. A COO⁻ symmetric stretch vibration causes the 1402 cm⁻¹ peak, and a very weak band at 1424 cm⁻¹ reflects CH₂ and CH₃ deformations; at 1444 cm⁻¹ CH₂ scissoring modes are found (traces 1 and 2 of Figure 3). The difference features visible in Figure 5 between 1410 and 1460 cm⁻¹ may be assigned to backbone vibrations of the DNA and/or changes in the protein. X-ray structure analyses indicate that DNA binding does not cause any significant distortion to the protein secondary structure relative to the free protein in solution but only causes rearrangement of some protein side chains (6, 11). Since the protein contributions to the 1424 cm⁻¹ band are very small, we conclude that vibrations of furanose residues are the main cause for the difference feature near 1420 cm⁻¹, whereas the peak near 1450 cm⁻¹ is mostly due to rearrangement of some protein side chains. The spectral effects might reflect either direct contacts of the furanose residues with protein side chains or conformational perturbations of the furanose rings. The difference feature near 1420 cm⁻¹ shows about 20% intensity change, which

implies that approximately two bases of each DNA strand undergo conformational perturbations (16, 26, 29).

Purine Environment. The crystal structure of Sac7d with the oligonucleotide octamer d(GCGATCGC)₂ (Figure 1) indicates two direct contacts between protein residue Ser31 and guanine G3 in the minor groove (6). Ser31 forms a hydrogen bond with the N2 amino group of the guanine base at position 3 (G3) that acts as the proton donor, and the indole N1H group of Trp24 forms a specific hydrogen bond to G3–N3 (6). A basically similar 3D structure is expected but still not available for the Sac7d–d(GAGGCGCCTC)₂ complex. In accordance with this expectation, prominent features in the difference spectra of the Sac7d–d(GAGGCGCCTC)₂ complex (Figure 5) are assignable to minor groove interaction of Sac7d with guanine N3.

Sac7d binding to guanine N2 and N3 causes changes of purine marker bands that result in the peak/trough features at $1342 (1349)/1354 (1363) \text{ cm}^{-1}$ (overlapping with the tryptophan Fermi doublet and the adenosine conformation marker) and 1571(1569)/1587 (1584) cm⁻¹ of the difference spectra (Figure 5). The band near 1355 cm⁻¹ (in D₂O) was assigned to the guanine ND_2 deformation vibration (34). The D₂O difference spectrum (Figure 5B) clearly exhibits a peak/ trough difference feature at 1349/1363 cm⁻¹; this feature is probably the spectral evidence for the contact of Ser31 with the N2 amino group of one or both guanines at positions 3 and/or 6 (see Discussion) in the minor groove. A similar feature but at higher wavenumber is expected to appear in the H₂O spectrum, but instead that peak/trough feature arises at lower wavenumber at about 1342/1354 cm⁻¹ (Figure 5A). A possible explanation is that the intense vibration of the N2 amino group of guanine (observed in D₂O near 1355 cm⁻¹) has in H₂O two weak components (Figure 3-2): the first one near 1364 cm⁻¹ (shoulder of the thymine 1376 cm⁻¹ band) and the second one near 1348 cm⁻¹ (shoulder of the adenine band near 1341 cm⁻¹). A tryptophan contribution of the Fermi doublet is not excluded by this consideration but, in any case, is not completely responsible for the observed difference feature because of the downshift upon deuteration. Components of the Fermi doublet are expected to form difference bands in H₂O solution near 1357 and 1340 cm^{-1} and in D₂O near 1350 and 1333 cm^{-1} (24).

Unexpected is the appearance of a difference feature with a peak at 1482 cm⁻¹ (in D₂O 1479 cm⁻¹) and a trough at 1494 cm⁻¹ (in D₂O 1488 cm⁻¹) (Figure 5). This difference feature is a purine ring vibration of guanine N7 which was several times observed as a result of major groove DNA binding of proteins (16-17, 35-36). In the Sac7dd(GCGATCGC)₂ system major groove binding is excluded by the X-ray structure analysis (6-8), and by extrapolation it is rather unlikely for the Sac7d-d(GAGGCGCCTC)2 complex. There are some possible explanations for the observed guanine N7 difference feature. First, Sac7d binding mediates an increase of the strength of the hydrogen bond of guanine N7 that is formed with a water molecule from the surroundings. Second, the guanine N7 ring vibration might be affected by the interactions of Sac7d with guanine ring N3 and/or N2 that are observed in the X-ray structure of the complex. Third, the B- to A-form transition of the DNA might cause a change in the guanine N7 vibration (32). The band near 1577 cm $^{-1}$ (in D₂O 1580 cm $^{-1}$) is probably altered because of hydrogen bond formation between guanine N3 and the indole N1H group of tryptophan that acts as proton donor.

Tryptophan. The Sac7d protein carries only one Trp residue (W24) that actively participates in binding of Sac7d to an 8-bp oligonucleotide (6) and thus facilitates the detection of key tryptophan markers in Sac7d–DNA complexes. Binding of Sac7d to d(GAGGCGCCTC)₂ results in spectral perturbations in tryptophan marker bands. In protein spectra, a Trp vibration called the W17 mode results in a band that is located between 883 and 871 cm⁻¹, and a Trp band called the W18 mode is located around 761 cm⁻¹. A tryptophan side chain in a hydrophilic environment gives rise to a very strong W18 band while in a hydrophobic environment the band becomes weaker (37).

In the spectrum of the Sac7d-d(GAGGCGCCTC)₂ complex, the W17 and W18 modes exhibit large spectral changes in comparison to the cases of the component spectra. However, the effects observed for the W17 and W18 modes show opposite tendencies in comparison to the expected ones. The interaction should increase the strength of the hydrogen bond and result in a downshift of the W17 band. The position of the W17 mode does not show this expected frequency shift; it is located at the same frequency (878 cm⁻¹) as that in the spectrum of free Sac7d (Figure 5A), but the intensity of the band increases by about 30%. This result is confirmed by the large intensity increase of the W17 mode in the D₂O spectrum; the band is observed at 859 cm⁻¹, in agreement with the 20 cm^{-1} shift described upon N1 deuteration (37). For the W18 mode an intensity increase is expected (37). To the contrary, the W18 mode does not increase its intensity but causes a very strong and sharp peak/trough feature at 762/756 (in D_2O 757/752) cm⁻¹ that indicates a downshift of the frequency. The changes observed here for the Sac7dd(GAGGCGCCTC)₂ system clearly indicate the participation of Trp24 in DNA binding despite of the differences in the behavior of the W17 and W18 modes in the lysozyme binding to inhibitors (37).

Sac7d Secondary Structure. In the difference spectra are no features that could be assigned to conformational changes of Sac7d upon complex formation. In the complex, amide III and amide I overlap with strong bands of DNA. Therefore, in some spectral regions it is difficult to discriminate between spectroscopic signals from protein or DNA. $H \rightarrow D$ exchange helps to assign at least the amide III band because it shifts from 1230–1310 cm $^{-1}$ in H_2O to the lower wavenumbers (900–1000 cm $^{-1}$) of the amide III' band where DNA bands do not overlap. However, comparison of the H_2O and D_2O difference spectra does not exhibit clear spectral changes in the regions of the amide III band. We conclude that Sac7d has the same conformation in the complex and free in solution. This is in agreement with NMR data (11) and the results of the X-ray crystal structure analyses (6).

DISCUSSION

Raman spectroscopy of the Sac7d—d(GAGGCGCCTC)₂ complex indicates significant perturbations to the B-form DNA backbone as a consequence of Sac7d binding to the DNA decamer d(GAGGCGCCTC)₂. Several nucleotides (G, C, A) change their conformation from C2'-endo/anti to C3'-endo/anti. Evidence for a partial transition from B- to A-form is found in several regions of the difference spectra.

Table 2: A- and B-Form DNA Conformation Markers in H₂O^a

			changes observed upon			
A-form			complex formation ^c			
	(cm^{-1})	assignment	(cm^{-1})		(cm^{-1})	
inc		G	667	inc		
dec	680	G		dec	686	
shift	787	C, bk	776	shift	782	
inc		C3'-endo	808	inc		
dec	829	C2'-endo		dec	823	
shift	1260	C	1247	shift	1262	
inc/shift	1318	G	1312	shift	1323	
inc/shift	1341	A	1332	shift	1342	
shift	1420	bk	1412	shift	1425	
shift	1489	G	1482	shift	1494	
shift	1578	G	1566	shift	1584	
	inc dec shift inc shift inc/shift shift shift	inc dec 680 shift 787 inc dec 829 shift 1260 inc/shift 1318 inc/shift 1341 shift 1420 shift 1489	B-form (cm ⁻¹) assignment	B-form compine changes compine compi	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	

^a Abbreviations: dec, decrease in intensity; inc, increase in intensity; shift, change in peak position; G, guanine; C, cytosine; A, adenine; bk, backbone. ^b Raman data from ref 32. ^c Raman difference features from Figure 5A.

Relevant spectral perturbations are summarized in Table 2. The phosphate band near 1092 cm⁻¹ does not change significantly upon complex formation. The 1092 cm⁻¹ band is commonly used as the intensity standard because of its invariance to effects of protein binding (38). By contrast, in complexes with minor groove binding proteins, slight broadenings of the phosphate band were observed that were interpreted to indicate changes in the local electrostatic environment (29). Since Sac7d binds in the minor groove of DNA octamers also for the Sac7d-d(GAGGCGCCTC)₂ system, minor changes of the 1092 cm⁻¹ band are not excluded. The identification of intensity differences is not possible in our experiments, since the 1092 cm⁻¹ band was used as an intensity standard for spectra normalization. However, broadening might occur and would yield a peak/ trough like feature. This was not observed and suggests a conservation of the local electrostatic environment of the phosphates upon minor groove binding of Sac7d.

High resolution structures exist for several Sac7d-DNA octamer complexes. This provides the possibility to analyze the available Raman spectra in the context of the wealth of known structural details. Crystal structures were resolved of Sac7d and Sso7d complexes with seven different octameric oligonucleotides. Several observations are in common. The DNA was found sharply kinked at the C2pG3 step in the Sac7d-(5'-GCGATCGC)₂ complex, at the A3pA4 step in the Sac7d-(5'-GTAATTAC)₂ complex (6), and for a T-G mismatched base pair at the T6pG7 step in the Sac7d-(5'-GCGATTGC)₂ complex (7), and the base at the 3' end of the intercalating site is always purine, with Trp 24 N1 forming a hydrogen bond to its N3 atom. Kinks were not observed at the GpC or the GpT steps, likely because of their more favorable base—base stacking energies (7). The sequence of d(GAGGCGCCTC)2 used for the present study does not contain TpG or ApA (TpT) steps.

The Raman results presented here show that in the Sac7d-d(GAGGCGCCTC)₂ complex one of the guanines serves as a binding site, and it is reasonable that the XpG steps of the d(GAGGCGCCTC)₂ sequence serve as intercalation sites. Potential candidates are the A2pG3, G3pG4, and C5pG6 steps. It has been suggested previously that a preferred site for protein side chain binding might be

connected with the base—base stacking energies (7). Up to now, to our knowledge base—base stacking energy calculations are not available that could be considered for a qualified discussion of the potential protein binding site in our system. In view of the size of d(GAGGCGCCTC)₂, the suggestion that the A2pG3 step could serve as a binding site implies the possibility that two Sac7d molecules bind to the oligonucleotide. However, gel electrophoresis of a complex formed in 1:1 molar stoichiometry did not provide any evidence in support of this assumption (data not shown).

The third base at the 3' end of the intercalation site is a thymine that forms a hydrogen bond with its O2 thymine oxygen to Arg43 (7). In the d(GAGGCGCCTC)₂ sequence studied here, the third base at the 3' ends of the two suggested intercalation sites is a cytosine. It is possible that Arg43 forms a hydrogen bond to cytosine O2, but the difference spectra do not provide feasible evidence for such a contact. It is not excluded from the data that Sac7d forms only two hydrogen bonds to guanine. The crystal structure of Sso7d-(5'-GTGATCGC)₂ shows that Arg43 NH1 is close to the third base (distance to T5-O2: 2.99 Å) and the fourth base (distances to C6-02 and C6-O4: 2.88 and 3.09 Å, respectively) at the 3' ends of the intercalation sites (7). In the case that Sac7d kinks d(GAGGCGCCTC)₂ at the C5pG6 step, thymine is the fourth base at the 3' end of the intercalation sites. The Sac7d Arg43 could be in a similar position as Arg43 of the Sso7d-(5'-GTGATCGC)₂ complex and form a hydrogen bond with T9O2. This could increase the probability of forming a complex with the kink at the C5pG6 step.

The structural adjustment of DNA to Sac7d binding occurs with little or no apparent consequences to the secondary structure of Sac7d, suggesting an induced fit mechanism with a more flexible DNA conformation. This is consistent with the highly stable secondary structure of Sac7d that was confirmed by both NMR (11) and crystallographic (6) studies.

DNA minor groove binding by Sac7d generates large conformational perturbations in the DNA, including a kink of approximately 60° and slight roll of roughly 10°, altogether making a bend of $\sim 70^{\circ}$ in the helical axis of d(GCGATCGC)₂. Many nucleotides that surround the wedge site adopt the less common C3'-endo (N-type) sugar puckers (6). The DNA Raman spectrum responds with extraordinarily large perturbations, which are consistent with the crystal structures of Sac7d/Sso7d-DNA complexes (6-8). These complexes exhibit unusual conformational features resembling those of A-DNA (as in an RNA duplex). The extent of the Raman changes is unusually high and varies from band to band. About 30% intensity changes relative to the parent bands were observed for Raman bands directly connected with nucleoside vibrations. The bands carrying information about base environments show $\sim 10-15\%$ relative intensity changes. A direct quantitative analysis of B- to A-DNA partial transitions is difficult because of the complexity of the Raman signals, where the relevant bands overlap with contributions from base-base interaction and changes in the surroundings of the bases. Nevertheless, from the extent of the intensity changes, we estimate that two to six bases resemble an A-DNA like conformation. The Raman spectroscopic data provide evidence for the involvement of guanine, adenine, and cytosine in Sac7d-d(GAGGCGCCTC)₂ complex formation, directly by serving as hydrogen bond donor or acceptor or indirectly by changing their conformation. There is strong spectroscopic evidence for intercalation; however, it is not possible to determine the intercalation site. Potential sites are the A2pG3, G3pG4, and C5pG6 steps. It should be mentioned that neither Sac7d nor Sso7d crystallize so far as complexes with d(GAGGCGCCTC)₂. Sac7d binding to more than one of the A2pG3, G3pG4, or C5pG6 sites and formation of heterogeneous mixtures of complexes would render crystallization difficult and explain this failure.

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